

Syntheses of New {Tetrakis(1-pyrazolyl)borato}samarium(III) Complexes and Their Temperature-Dependent Exchange Motions of All Pyrazolyl Groups

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As the first unambiguous syntheses and characterization of the {tetrakis(1-pyrazolyl)borato}lanthanide(III) complexes, stable eight-coordinate (β -ketoenolato)samariums [$\text{Sm}\{\eta^3\text{-B}(\text{pz})_4\}_2(\beta\text{-ketoenolato})$] (pz = 1-pyrazolyl group) were prepared, and novel stereochemically non-rigid temperature-dependent motions of the $\text{B}(\text{pz})_4$ ligands were observed on their solution-state ^1H -NMR spectra, showing the spectroscopic equivalence of coordinated and uncoordinated pyrazolyl groups at high temperatures.

Beneficial characteristics of bis(pentamethylcyclopentadienyl) ($\text{Cp}^*\text{2}$) supporting ligation have been realized in the rapid growth of lanthanide chemistry during the last two decades.¹ As one comparable class to the uninegative cyclopentadienyls with 6e-donative facial coordination, there are η^3 -polypyrazolylborate ligands, formulated as $[\text{BR}(\text{pz})_3]^-$ (pz = 1-pyrazolyl group). The polypyrazolylborate ligands, especially tetrakis(1-pyrazolyl)borate $[\text{B}(\text{pz})_4]^-$ with a more widely delocalized π -conjugated system, have been shown to confer considerable stability on a variety of ruthenium and some other transition-metal organometallic complexes, and incentive for studies on these transition-metal complexes was derived also from some successful modification of electronic and steric properties of the ligands upon their proper substituent introduction.² Concerning lanthanide(III) ions, the complexes of $[\text{M}\{\text{BR}(\text{pz})_3\}_2\text{Cl}(\text{L})_n]$ (L , monodentate neutral ligands), which are regarded as typical precursors to the polypyrazolylborate lanthanide(III) chemistry, have been prepared and studied fairly well,³ but seemed to be limited to the $\text{BH}(\text{pz})_3$ series. The corresponding $\text{B}(\text{pz})_4$ complexes have not been isolated so far from the reactions with lanthanide(III) chlorides $\text{MCl}_3 \cdot 6\text{H}_2\text{O}$.⁴ In this study, the potassium salt $\text{K}[\text{B}(\text{pz})_4]$ was allowed to react with samarium(III) triflate $\text{Sm}(\text{OTf})_3$,⁵ and subsequent treatments with sodium β -ketoenolates gave the stable complexes of $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\beta\text{-ketoenolato})]$,⁶ for the first unambiguous syntheses and characterization of {tetrakis(1-pyrazolyl)borato}lanthanide(III) complexes. Moreover, novel stereochemically non-rigid temperature-dependent motions of the $\text{B}(\text{pz})_4$ also were found in these complexes, on the basis of solution-state ^1H -NMR spectroscopy.

The triflate $\text{Sm}(\text{OTf})_3$ in THF was treated with twofold moles of the $\text{K}[\text{B}(\text{pz})_4]$ salt to afford white microcrystals, probably assignable to $\text{Sm}(\text{Bpz}_4)_2(\text{OTf})(\text{H}_2\text{O})_n$. Decomposition of the salt was not observed. Their subsequent reactions were performed with some sodium β -ketoenolates, involving those of pentane-2,4-dionate ($\text{Na}[\text{acac}]$), salicylaldehyde ($\text{Na}[\text{sal}]$), 2,2,6,6-tetramethylheptane-3,5-dionate, and 3-methylpentane-2,4-dionate, and in high yields the $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\beta\text{-ketoenolato})]$ complexes were isolated.⁷

X-ray structural analysis was carried out for colorless prismatic crystals of $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\text{sal})]$, and confirmed its monomeric eight-coordinate structure with two $\eta^3\text{-B}(\text{pz})_4$ chelates, free from conceivable water-molecule coordination (Figure 1).⁸ The Sm-N coordinate-bond lengths ranged from 2.523 to 2.602 Å, with an

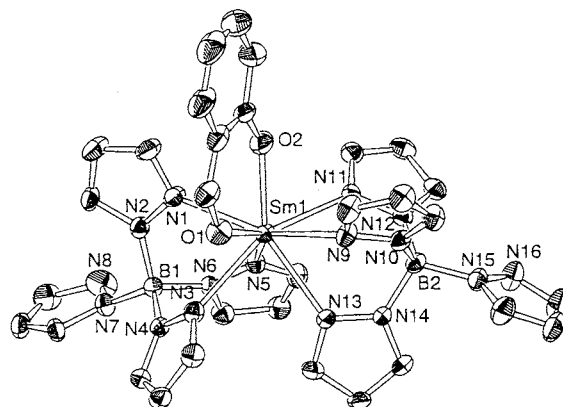


Figure 1. ORTEP view of $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\text{sal})]$ showing 30% probability thermal ellipsoids and atom labels for only important atoms. Selected distances (Å) and angles (deg); Sm(1)-O(1) 2.411(5), Sm(1)-O(2) 2.278(5), Sm(1)-B(1) 3.741(8), Sm(1)-B(2) 3.725(8), Sm(1)-N(1,3,5)_{av} 2.562, Sm(1)-N(9,11,13)_{av} 2.560, O(1)-Sm(1)-O(2) 73.0(2), B(1)-Sm(1)-B(2) 128.6(2).

average of 2.561 Å. Furthermore, the complex was found to involve two nearly planar faces of O(1)-N(3)-N(9)-N(13) and O(2)-N(1)-N(5)-N(11), and the dihedral angle between these two faces was only 1.21°. Accordingly, the coordination structure of the complex in the solid state was best described as a distorted square-antiprismatic geometry.

On the ^1H -NMR spectra of $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\beta\text{-ketoenolato})]$ in the present study, β -ketoenolato protons showed sharp signal patterns satisfactorily,⁹ whereas each of the $\text{B}(\text{pz})_4$ ligands gave two or three significantly broad signals at room temperature. Observation of variable-temperature ^1H -NMR revealed stereochemically non-rigid temperature-dependent motions of the ligands. As one example, at -80 °C, two $\text{B}(\text{pz})_4$ ligands in $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\text{acac})]$ exhibited three NMR signals at δ 2.78, 5.64, and 9.08 for six spectroscopically equivalent coordinated pyrazolyl groups, and three ones at δ 6.70, 8.43, and 8.52 for two uncoordinated groups. On the other hand at 100 °C, the $\text{B}(\text{pz})_4$ ligands gave three signals at δ 5.38, 5.90, and 8.34 for eight pyrazolyl groups, indicating spectroscopical equivalence of the four pyrazolyl groups in the respective $\text{B}(\text{pz})_4$ ligands, due to fast exchange between coordinated and uncoordinated groups.¹⁰ In spite of the η^3 -coordination-favoring chelate frame of ligating $\text{B}(\text{pz})_4$ ligands, strengthened also by the electrostatic attraction of the negative charge on boron atoms with samariums, each of the Sm-N coordinate bonds was found to be still weak, enough to bring about dissociation of one coordinated pyrazolyl groups, i.e. formation of short-lived $\eta^2\text{-B}(\text{pz})_4$ transients, probably followed by inversion of the six-membered Sm-(N-N)₂-B ring and further by coordination of the previously uncoordinated pyrazolyl group.

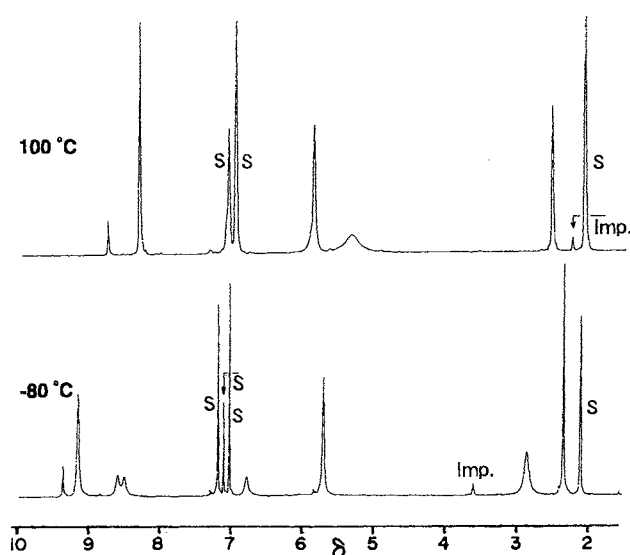


Figure 2. ^1H -NMR of $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\text{acac})]$ in $\text{C}_6\text{D}_5\text{CD}_3$. S, solvent; Imp., impurity.

In summary, we prepared the {tetrakis(1-pyrazolyl)borato}lanthanides(III) of $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\beta\text{-ketoenolato})]$, which showed novel stereochemically non-rigid temperature-dependent motions of the $\text{B}(\text{pz})_4$ ligands on their ^1H -NMR spectra, involving the fast exchange of coordinated and uncoordinated pyrazolyl groups at high temperatures.

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- Crystal data; $(\text{C}_{31}\text{H}_{29}\text{B}_2\text{N}_{16}\text{O}_2\text{Sm})$, $M = 829.70$; triclinic, space group $P\bar{1}(\text{No.}2)$, $a = 14.035(2)$, $b = 27.143(2)$, $c = 9.485(2)$ Å, $\alpha = 98.975(10)^\circ$, $\beta = 91.23(1)^\circ$, $\gamma = 102.161(8)^\circ$, $V = 3483.7(8)$ Å³, $Z = 4$, $D_{\text{calc}} = 1.582$ g/cm³, $\mu(\text{MoK}\alpha) = 17.45$ cm⁻¹. Data collected at 296 K with a Rigaku AFC-7S diffractometer (Mo radiation, $\lambda = 0.71069$ Å; $2\theta_{\text{max}} = 55.0^\circ$); $R = 0.045$ and $R_w = 0.048$ for 10016 observations ($I > 3.00\sigma(I)$). $\text{Sm}(1)\text{-N}(1)$, 2.544(6); $\text{Sm}(1)\text{-N}(3)$, 2.544(6); $\text{Sm}(1)\text{-N}(5)$, 2.599(5); $\text{Sm}(1)\text{-N}(9)$, 2.602(6); $\text{Sm}(1)\text{-N}(11)$, 2.523(6); $\text{Sm}(1)\text{-N}(13)$, 2.556(5) Å.
- NMR signals of acac in $[\text{Sm}\{\text{B}(\text{pz})_4\}_2(\text{acac})]$ ($\text{C}_6\text{D}_5\text{CD}_3$); δ 2.27 and 9.29 at -80°C ; δ 2.56 and 8.78 at 100°C .
- Similar stereochemically non-rigid temperature-dependent dynamics of the ligating $\text{B}(\text{pz})_4$ ligands, which induce ^1H -NMR spectroscopic equivalence of their coordinated and uncoordinated pyrazolyl groups at high temperatures, have been described to date for some d^8 transition-metal complexes $\{\text{Pd}(\text{II}), \text{Pt}(\text{II}), \text{and Rh}(\text{I})\}$ and for a few of $[\text{M}\{\text{B}(\text{pz})_4\}_2]$ -type compounds of Group 14 typical-elements $\{\text{M} = \text{Sn}(\text{II}) \text{ and } \text{Pb}(\text{II})\}$. Mutual interchange between η^2 and η^3 -coordination modes of the $\text{B}(\text{pz})_4$ ligands and inversion of boat-type six-membered $\text{M}(\text{N-N})_2\text{-B}$ rings in the η^2 -species have been regarded as essential key steps, affording the spectroscopic equivalence of the coordinated and uncoordinated pyrazolyl groups. See for example; L. E. Manzer and P. Z. Meakin, *Inorg. Chem.*, **15**, 3117 (1976); M. Cocivera, G. Ferguson, F. J. Lalor, and P. Szczecinski, *Organometallics*, **1**, 1139 (1982); D. L. Reger, S. J. Knox, M. F. Huff, A. Rheingold, and B. S. Haggerty, *Inorg. Chem.*, **30**, 1754 (1991); D. L. Reger, M. F. Huff, A. L. Rheingold, and B. S. Haggerty, *J. Am. Chem. Soc.*, **114**, 579 (1992); M. Onishi, K. Hiraki, M. Shironita, Y. Yamaguchi, and S. Nakagawa, *Bull. Chem. Soc. Jpn.*, **53**, 961 (1980). Furthermore, it is noted that zinc and thallium compounds, $[\text{Zn}\{\text{B}(\text{pz})_4\}_2]$ and $[\text{TlR}_2\{\text{B}(\text{pz})_4\}]$ have shown similar NMR spectroscopic equivalence of all four pyrazolyl groups. However, η^3 -coordinate bonding character of the $\text{B}(\text{pz})_4$ ligands has not been ascertained for these compounds in the solution state even at low temperatures, and accordingly these equivalence seems to be attributable to their ionic bonding-character between the metal atoms and the $\text{B}(\text{pz})_4$ ligands. S. Trofimenko, *Acc. Chem. Res.*, **4**, 17 (1971); M. Onishi, K. Hiraki, and S. Nakagawa, *Bull. Chem. Soc. Jpn.*, **53**, 1459 (1980).